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(vi) $\text{HCOOC}(\text{CH}_3)_3$ is an isomer of X For the ^1H NMR spectrum of this isomer, deduce the total number of peaks (excluding the TMS peak at 0 ppm) and the ratio of peak areas For each peak, deduce whether it is a singlet, doublet, triplet, quartet or shows a more complex splitting pattern [3]

Number of peaks (excluding the TMS peak at 0 ppm):

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to (aluminium hydroxide) $\frac{3}{03}$ mol OH^- ions for neutralization / neutralizes $2\text{H}^+ / 02$ mol acid as compared to $3\text{H}^+ / 03$ mol acid; [1] Do not accept aluminium hydroxide can neutralize more acid

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